

Basic Principles of Protein Structures

ECS129
PATRICE KOEHL

Proteins

Proteins: The Molecule of Life

Proteins: Building Blocks

Proteins: Secondary Structures

Proteins: Tertiary and Quaternary Structure

Proteins: Geometry



Proteins

Proteins: The Molecule of Life

Proteins: Building Blocks

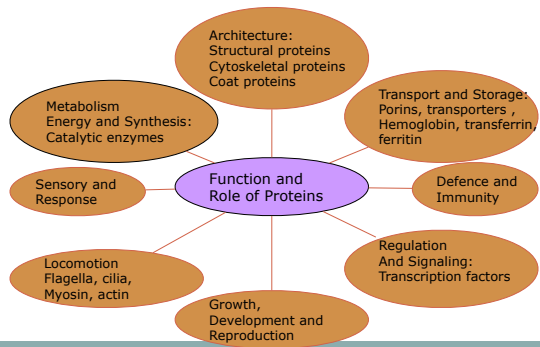
Proteins: Secondary Structures

Proteins: Tertiary and Quaternary Structure

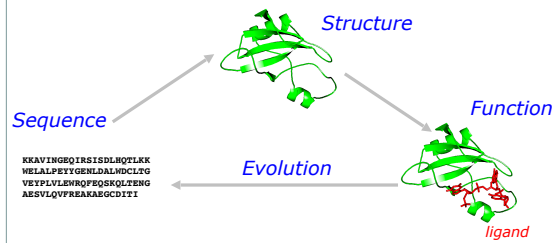
Proteins: Geometry



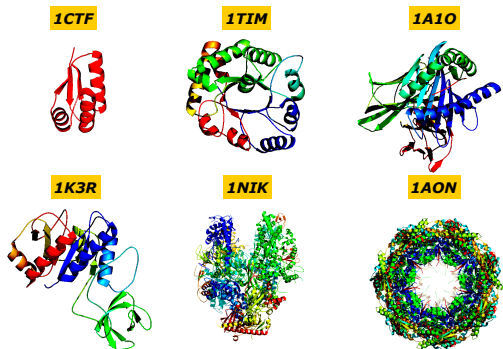
Why Proteins?



The Protein Cycle

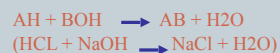


Protein Structure Diversity



Review of Acid-Base Chemistry

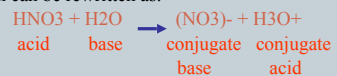
Acid plus base makes water plus a salt:



The chemical dissociation of nitric acid is:



Which can be rewritten as:



Review of Acid-Base Chemistry

pH is a measure of how acidic or alkaline (basic) a solution is. The pH of a solution is the negative log of the hydrogen ion concentration.

$$pH = -\log([H^+])$$

$$pOH = -\log([OH^-])$$

$$pH + pOH = 14$$

	[H ⁺]	pH	pOH	[OH ⁻]
Strong base	10 ⁻¹⁴	14	0	1
Base	10 ⁻¹²	12	2	10 ⁻²
Weak base	10 ⁻⁹	9	5	10 ⁻⁵
Neutral	10 ⁻⁷	7	7	10 ⁻⁷
Weak acid	10 ⁻⁴	4	10	10 ⁻¹⁰
Acid	10 ⁻²	2	12	10 ⁻¹²
Strong acid	1	0	14	10 ⁻¹⁴

Review of Acid-Base Chemistry

Dissociation of a weak acid:



Equilibrium constant:

$$K_A = \frac{[H^+][A^-]}{[HA]}$$

$$pK_A = -\log(K_A)$$

Dissociation of a weak base:



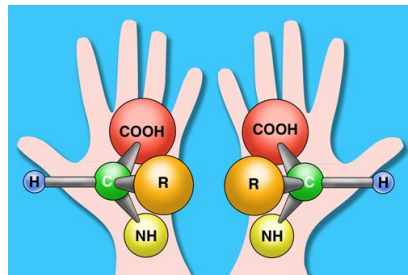
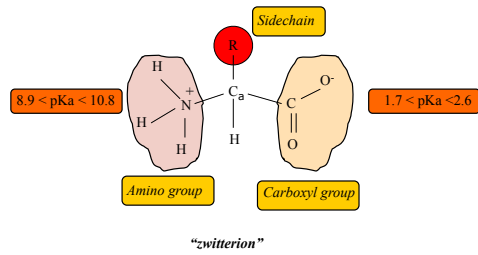
$$K_B = \frac{[B^+][OH^-]}{[BOH]}$$

$$pK_B = -\log(K_B)$$

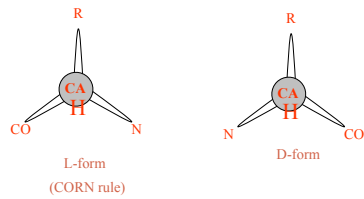
For an (acid,base) pair:

$$pK_A + pK_B = 14$$

The Basic Block: Amino Acid



Amino Acid Chirality



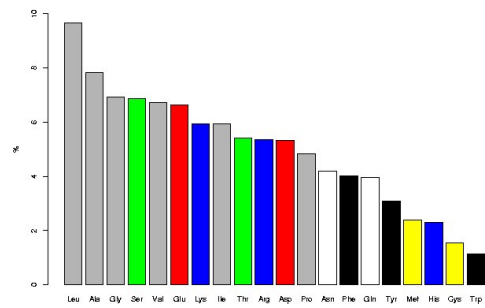
Amino acids in proteins are in the L-form

Threonine and Isoleucine have a second optical center which is also identical in all natural amino acids.

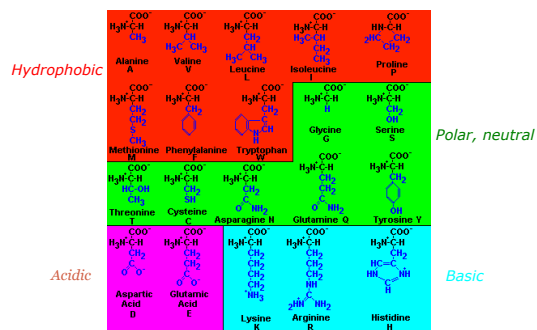
The 20 amino acids

1-letter	3-letter	Amino acid	1-letter	3-letter	Amino Acid
A	Ala	Alanine	M	Met	Methionin
C	Cys	Cysteine	N	Asn	Asparagine
D	Asp	Aspartic Acid	P	Pro	Proline
E	Glu	Glutamic Acid	Q	Gln	Glutamine
F	Phe	Phenylalanine	R	Arg	Arginine
G	Gly	Glycine	S	Ser	Serine
H	His	Histidine	T	Thr	Threonin
I	Ile	Isoleucine	V	Val	Valine
K	Lys	Lysine	W	Trp	Tryptophan
L	Leu	Leucine	Y	Tyr	Tyrosine

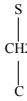
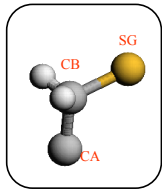
Amino Acids: Usage



The 20 amino acids



Polar Amino acids: Cysteine



Names: Cys, C

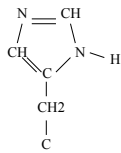
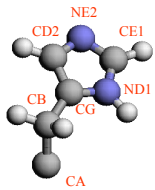
Occurrence: 1.8 %

pKa sidechain: 8.3

Can form disulphide bridges
in proteins



Polar Amino acids: Histidine

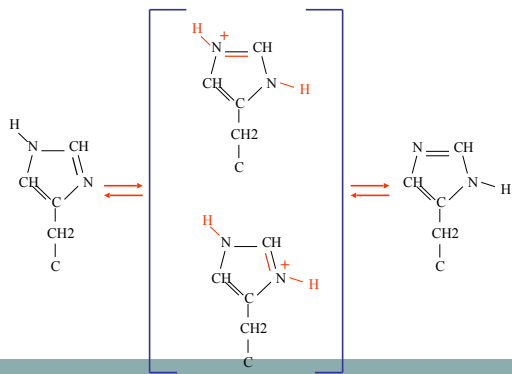


Name: His, H

Occurrence: 2.2 %

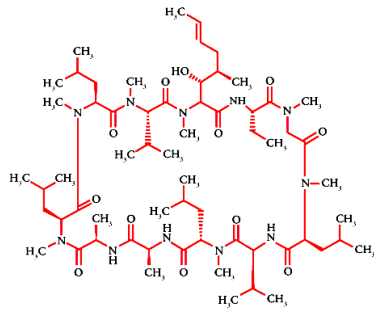
pKa sidechain: 6.04

Different ionic states of Histidine



Unusual Amino Acids: Cyclosporin

Correct!!



<http://www.cellsignal.com/products/9973.html>

Structural Bioinformatics: Proteins

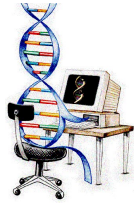
Proteins: The Molecule of Life

Proteins: Building Blocks

Proteins: Secondary Structures

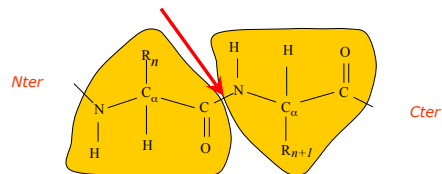
Proteins: Tertiary and Quaternary Structure

Proteins: Geometry

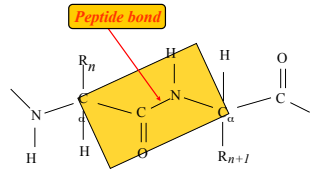


The Protein: A polymer of Amino acids

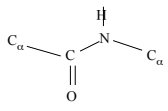
Peptide bond



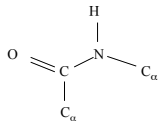
The Peptide Bond



The peptide bond is planar



Conformation "Trans"



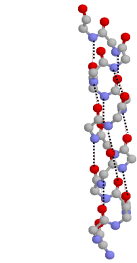
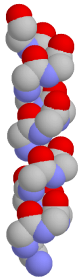
Conformation "Cis"

Helices

Cter

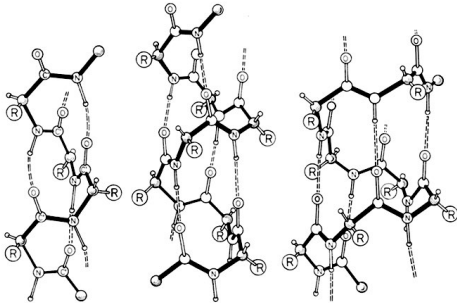


Nter



Hydrogen bonds: O (i) \leftrightarrow N (i+4)

Helices



3₁₀ helix

α-helix (4₁₂)

π-helix (5₁₆)

Helices

3_{10} helix

"Thin"; 3.0 residues /turn; ~ 4 % of all helices

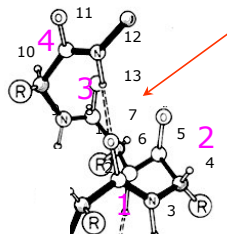
π -helix (5_{16})

"Fat"; 4.2 residues /turn; instable

α -helix (4_{13})

"Right"; 3.6 residues /turn; 5.4 Å /turn; most helices

Identify Helix Type



1. Find one hydrogen bond loop

2. Count number of residues
(by number of C atoms in the loop).
Here :

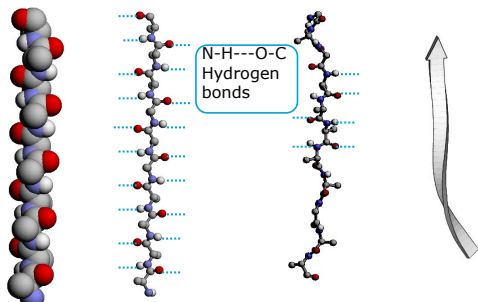
4

3. Count number of atoms in the loop
(including first O and last H). Here:

13

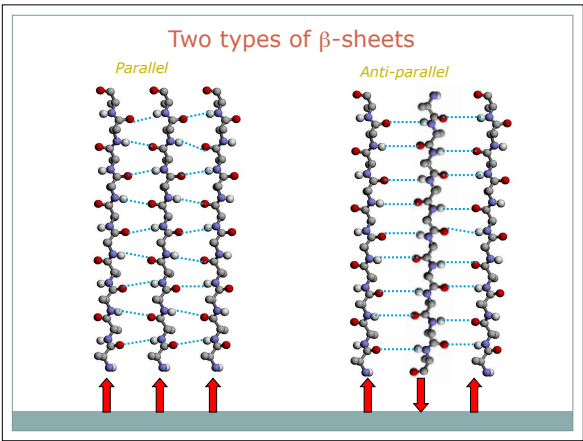
4_{13} helix = α -helix

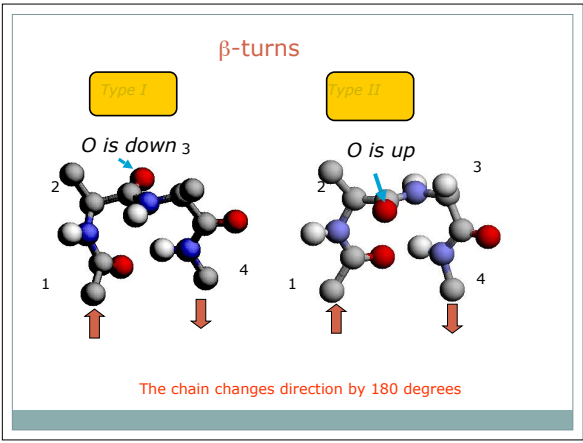
The β -strand



Extended chain is flat

"Real β -strand is twisted"

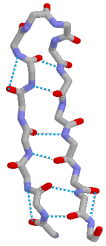




Favorable /Unfavorable Residues In Turns

Turn	1	2	3	4
I	Asp, Asn, Ser, Cys	Pro	Pro	Gly
II	Asp, Asn, Ser, Cys	Pro	Gly, Asn	Gly

The β -hairpin



Structural Bioinformatics: Proteins

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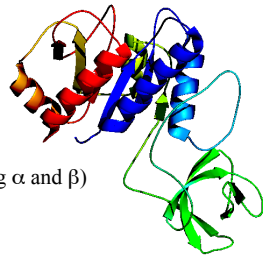
Proteins: Tertiary and Quaternary Structure

Proteins: Geometry



Protein Tertiary Structure

- All α proteins
- All β proteins
- Alpha and beta proteins:
 - α/β proteins (alternating α and β)
 - $\alpha + \beta$ proteins



All-Alpha topologies

- The lone helix



*Glucagon (hormone involved
in regulating sugar metabolism)*
PDB code: 1GCN

- The helix-turn-helix motif

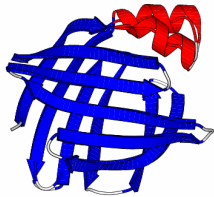


*ROP: RNA-binding
Protein*
PDB code: 1ROP

*The 2 helices
are twisted*

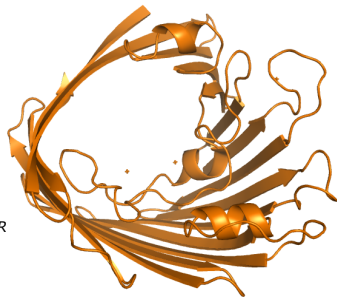
All Beta Topology

Beta sandwiches:



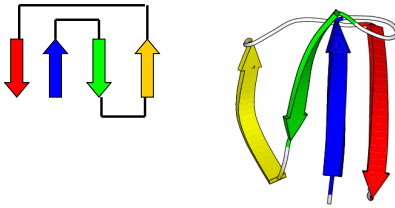
Fatty acid binding protein
PDB code: 1IFB

Closed Beta Barrel



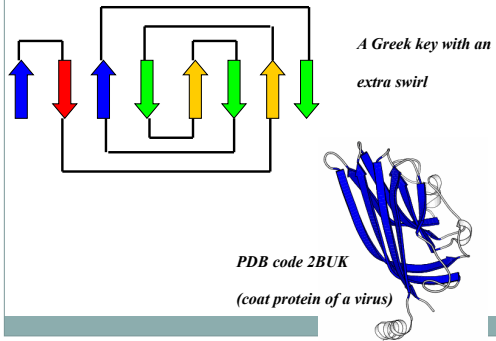
PDB file: 2POR

The Greek Key Topology

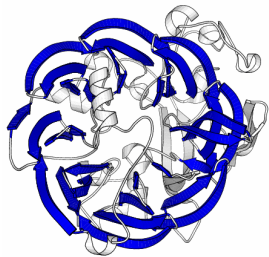


Folds including the Greek key topology include 4 to 13 strands.

The Jellyroll Topology



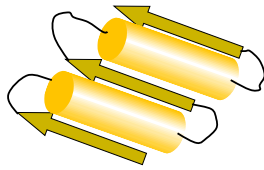
The Beta Propellor



*Eight-plated propellor:
Each plate is a four-stranded
anti-parallel sheet
PDB code 4AAH*

Alpha- Beta Topology

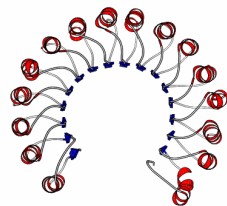
The Rossman fold:



Alternate beta / alpha motif

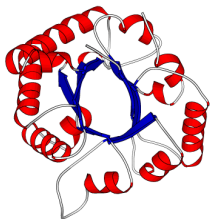
Always right handed

The Horseshoe



PDB code: 2BNH

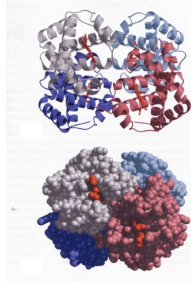
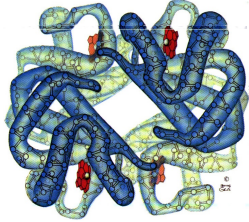
The alpha/beta barrel



In a succession of alpha/beta motifs, if the first strand connects to the last, then the structure resembles a Barrel.

PDB code : 1TIM

Quaternary Structures Assemblies of Protein Chains



Hemoglobin - 4 chains:
2- α chain, 2- β chain
(Heme- four iron groups)

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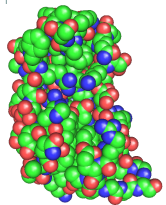
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Structure

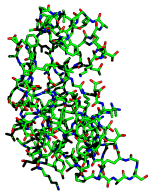
Proteins: Geometry



Protein Structure Representation



CPK: hard sphere model



Ball-and-stick



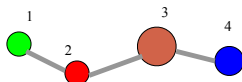
Cartoon

Degrees of Freedom in Proteins

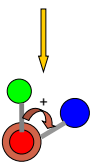
Bond length



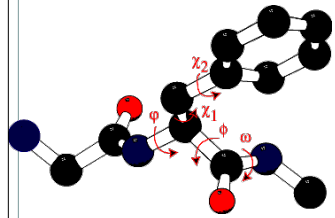
Dihedral angle



Bond angle



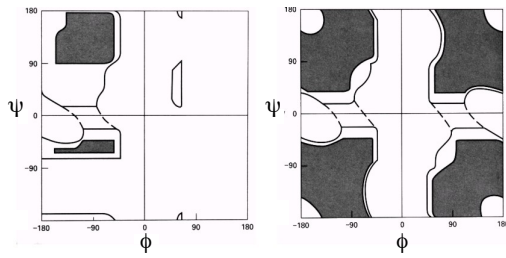
Protein Structure: Variables



Backbone: 3 angles per residue : ϕ , ψ and ω

Sidechain: 1 to 7 angles, χ_i ; each χ has 3 favored values: 60° , -60° , 180° .

Ramachandran Plots



All residues, but glycine

Glycine

Acta Cryst. (2002). D58, 768-776

What have we learnt?

- All proteins are polymers built up from 20 amino acids.
- All 20 amino acids have a similar structure: they all have a main-chain, consisting of an amino group and an acidic group, attached to a central carbon, named CA; the remaining atoms form the side-chain, that can be hydrophobic, polar or charged (acid or basic).
- The conformation of the backbone of amino acids is restricted, except for glycine that does not have a sidechain.
- There are 3 main graphical representations of proteins: space-filling, wireframe and cartoon.

What have we learnt?

- There are 3 major types of secondary structures: α -helices, β -sheets and β -turns.
- Most helices are α -helices, stabilized through a network of CO (i) --- HN (i+4) hydrogen bonds
- There are two types of β -sheets: parallel and anti-parallel
- β -turns correspond to 180 change in the backbone direction.

What have we learnt?

- There are three main classes of proteins: all Alpha, all Beta and Alpha + Beta. The latter can be divided in two, considering the alternating alpha/beta proteins as defining their own class.
- Bundles are common alpha-proteins
- Common beta folds include the greek key and the sandwiches. Immuno-globulins adopt a beta fold.
- The Rossman fold (alternating alpha/beta) is a common motif in proteins. It is found in the horseshoe, as well as in the TIM fold.